

Supplementary Materials

In Vitro Screening of Anti-Viral and Virucidal Effects Against SARS-Cov-2 by Phenylamino-Phenoxy-Quinoline Derivatives: An Integrative Computational Approach

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These figures depict the two-dimensional binding mode and intricate binding interactions taking place in the active site of the Main Protease (M^{Pro}). The results of molecular docking showcase the binding interactions of quinoline derivatives **1(a-d)**, **2(a-d)**, and **3(a-d)**, in addition to commercially accessible medications (ritonavir, ensitrevir, and nirmatrelvir). The analysis of these interactions is conducted within the binding pocket, defined within a radius of 3.0 Å from M^{Pro}, as detailed below.

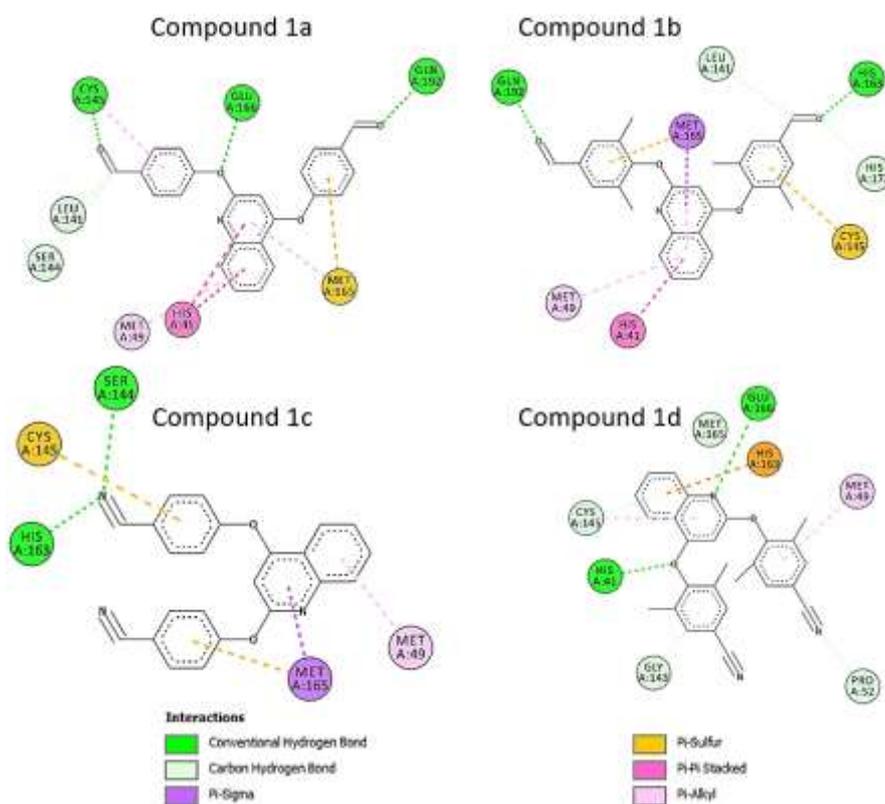


Figure S1 2-dimensional interaction diagram between compound **1(a-d)** and M^{Pro}.

For **2,4-di-(4'-formylphenoxy)-quinoline (1a)**, the amino acid residues of M^{Pro} within a 5.0 Å radius included HIS41, MET49, PRO52, TYR54, CYS145, HIS164, MET165, GLU166, LEU167, PRO168, ASP187, ARG188, GLN189, THR190, and GLN192. The conventional hydrogen bonding interactions between M^{Pro} and **1a** were observed, such as those between the H atom of the NH₂ group in CYS145 and the O atom of the C=O group in **1a**, between the H atom of the NH₂ group in GLN192 and the O atom of the C=O group in **1a**, and between the H atom of the NH₂ group in GLU166 and the O atom of **1a**. Additionally, three carbon hydrogen bonds were formed with LEU141, SER144, and ALA191, and one π -Sulfur interaction was displayed with MET165, and one π -Alkyl interaction was observed with MET49. In addition, one hydrophobic interaction was performed using π - π stacking between HIS41 and **1a**.

For **2,4-di-(2',6'-Dimethyl-4'-formylphenoxy)-quinoline (1b)**, the amino acid residues of M^{Pro} within a 5.0 Å radius included HIS41, MET49, PRO52, TYR54, CYS145, HIS164, MET165, GLU166, LEU167, PRO168, ASP187, ARG188, GLN189, THR190, and GLN192. The conventional hydrogen bonding interactions between M^{Pro} and **1b** were demonstrated, such as those between the H atom of the NH₂ group in HIS163 and the O atom of the C=O group in **1b**, and between the H atom of the NH₂ group in GLN192 and the O atom of the C=O group in **1b**. Two carbon hydrogen bonds were formed with LEU141 and HIS172, one π -Sulfur interaction was observed with CYS145, and one π -Alkyl interaction was observed with MET49. In addition, one hydrophobic interaction was performed using π - π stacking between HIS41 and **1b**.

For **2,4-di-(4'-cyanophenoxy)-quinoline (1c)**, the amino acid residues of M^{Pro} within a 5.0 Å radius included HIS41, MET49, PRO52, TYR54, CYS145, HIS164, MET165, GLU166, LEU167, PRO168, ASP187, ARG188, GLN189, THR190, and GLN192. The conventional hydrogen bonding interactions between M^{Pro} and **1c** were observed, such as those between the H atom of the OH group in SER144 and the N atom of the CN group in **1c** and between the H atom of the NH₂ group in HIS163 and the N atom of the CN group in **1c**. Moreover, two π -Sulfur interactions were displayed with CYS145 and MET165, and one π -Alkyl interaction was observed with MET49.

For **2,4-di-(2',6'-Dimethyl-4'-cyanophenoxy)-quinoline (1d)**, the amino acid residues of M^{Pro} within a 5.0 Å radius included HIS41, MET49, PRO52, TYR54, CYS145, HIS164, MET165, GLU166, LEU167, PRO168, ASP187, ARG188, GLN189, THR190, and GLN192. The conventional hydrogen bonding interactions between M^{Pro} and **1d** were demonstrated, such as those between the H atom of the NH₂ group in HIS41 and the N atom of **1d** and between the H atom of the NH₂ group in GLU166 and the O atom of **1d**. Two carbon hydrogen bonds were formed with PRO52 and MET165, two π -Donor hydrogen bond interactions were observed with GLY143, CYS145, and one π -Alkyl interaction was observed with MET49.

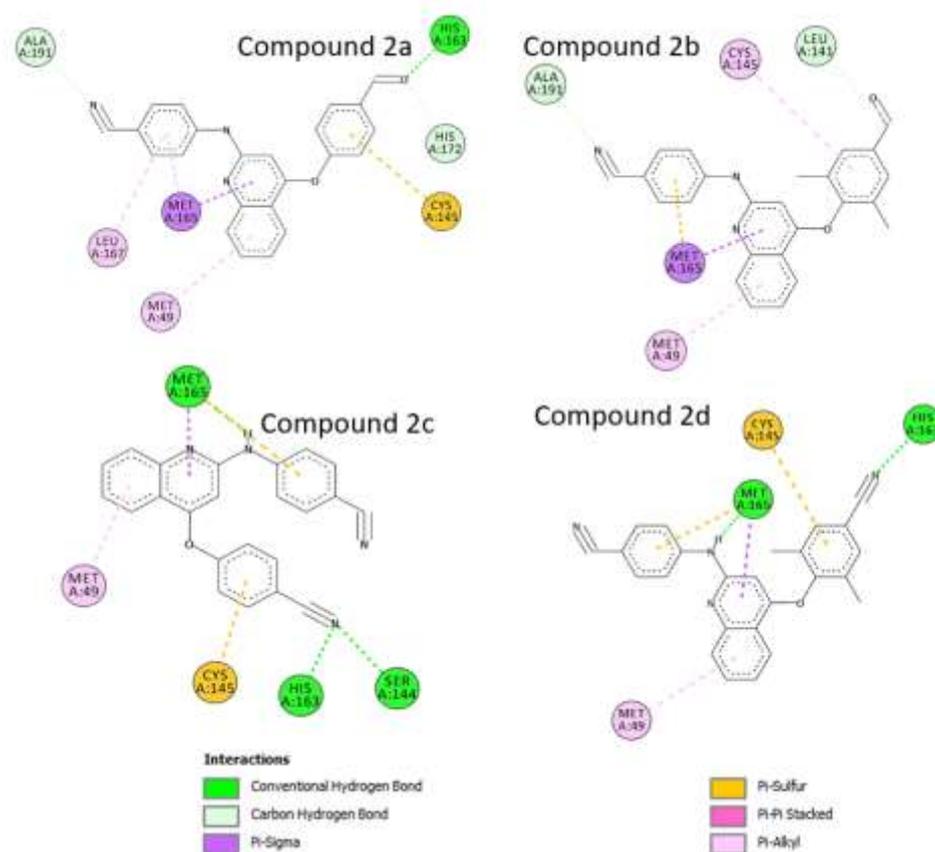


Figure S2 2-dimensional interaction diagram between compound **2(a-d)** and M^{pro} .

For **4-(4'-formylphenoxy)-2-(4''-cyanophenyl)-aminoquinoline (2a)**, the amino acid residues of M^{pro} within a 5.0 Å radius included HIS41, MET49, PRO52, TYR54, CYS145, HIS164, MET165, GLU166, LEU167, PRO168, ASP187, ARG188, GLN189, THR190, and GLN192. The conventional hydrogen bonding interactions between M^{pro} and **2a** were observed, such as those between the H atom of the NH_2 group in HIS163 and the O atom of the $\text{C}=\text{O}$ group in **2a**. Two carbon hydrogen bonds were formed with HIS172 and ALA191, and one π -Sulfur interaction was displayed with CYS145, along with two π -Alkyl interactions observed with MET49 and LEU167.

For **4-(2',6'-Dimethyl-4'-formylphenoxy)-2-(4''-cyanophenyl)-aminoquinoline (2b)**, the amino acid residues of M^{pro} within a 5.0 Å radius included HIS41, MET49, PRO52, TYR54, CYS145, HIS164, MET165, GLU166, LEU167, PRO168, ASP187, ARG188, GLN189, THR190, and GLN192. Two carbon hydrogen bonds were formed with LEU141 and ALA191, one π -Sulfur interaction was observed with MET165, and two π -Alkyl interactions were observed with MET49 and CYS145.

For **4-(4'-cyanophenoxy)-2-(4''-cyanophenyl)-aminoquinoline (2c)**, the amino acid residues of M^{pro} within a 5.0 Å radius included HIS41, MET49, PRO52, TYR54, CYS145, HIS164, MET165, GLU166, LEU167, PRO168, ASP187, ARG188, GLN189, THR190, and GLN192. The conventional hydrogen bonding interactions between M^{pro} and **2c** were observed, such as those between the H atom of the OH group in SER144 and the N atom of the CN group in **2c**, between the H atom of the NH_2 group in HIS163 and the N atom of the CN group in **2c**, and those between the S atom of the SH group

in MET165 and the H atom of the NH₂ group in **2c**. Moreover, one π -Sulfur interaction was displayed with CYS145, and one π -Alkyl interaction was observed with MET49.

For **4-(2',6'-Dimethyl-4'-cyanophenoxy)-2-(4''-cyanophenyl)-aminoquinoline (2d)**, the amino acid residues of M^{PRO} within a 5.0 Å radius included HIS41, MET49, PRO52, TYR54, CYS145, HIS164, MET165, GLU166, LEU167, PRO168, ASP187, ARG188, GLN189, THR190, and GLN192. The conventional hydrogen bonding interactions between M^{PRO} and **2d** were demonstrated, such as those between the H atom of the NH₂ group in HIS163 and the N atom of the CN group in **2d** and those between the S atom of the SH group in MET165 and the H atom of the NH₂ group in **2d**. Two π -Sulfur interactions were observed with CYS145, MET165, and one π -Alkyl interaction was observed with MET49.

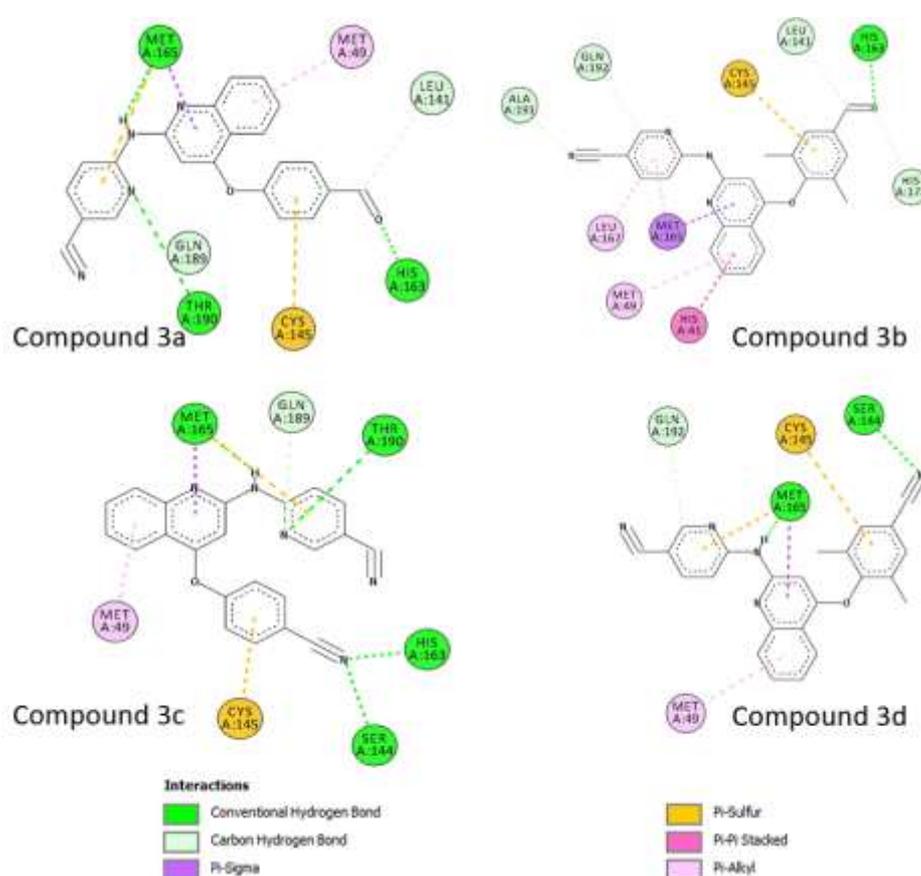


Figure S3 2-dimensional interaction diagram between compound **3(a-d)** and M^{PRO}.

For **4-(4'-formylphenoxy)-2-(5''-cyanopyridin-2''ylamino)quinoline (3a)**, the amino acid residues of M^{PRO} within a 5.0 Å radius included HIS41, MET49, PRO52, TYR54, CYS145, HIS164, MET165, GLU166, LEU167, PRO168, ASP187, ARG188, GLN189, THR190, and GLN192. The conventional hydrogen bonding interactions between M^{PRO} and **3a** were observed, such as those between the H atom of the NH₂ group in HIS163 and the O atom of the C=O group in **3a**, between the S atom of the SH group in MET165 and the H atom of the NH₂ group in **3a**, and between the H atom of the NH₂ group in THR190 and the N atom of the NH₂ group in **3a**. Additionally, two carbon hydrogen

bonds were formed with LEU141 and GLN189, and interactions included one π -Sulfur interaction with CYS145 and one π -Alkyl interaction with MET49.

For **4-(2',6'-Dimethyl-4'-formylphenoxy)-2-(5''-cyanopyridin-2''ylamino)quinoline (3b)**, the amino acid residues of M^{Pro} within a 5.0 Å radius included HIS41, MET49, PRO52, TYR54, CYS145, HIS164, MET165, GLU166, LEU167, PRO168, ASP187, ARG188, GLN189, THR190, and GLN192. The conventional hydrogen bonding interactions between M^{Pro} and **3b** were observed, such as those between the H atom of the NH₂ group in HIS163 and the O atom of the C=O group in **3b**. In addition, four carbon hydrogen bonds were formed with LEU141, HIS172, ALA191, and GLN192, and interactions included one π -Sulfur interaction with CYS145 and two π -Alkyl interactions with MET49 and LEU167. Furthermore, a π - π stacking interaction between the HIS41 residue and the benzene ring of **3b** was observed.

For **4-(4'-cyanophenoxy)-2-(5''-cyanopyridin-2''ylamino)quinoline (3c)**, the amino acid residues of M^{Pro} within a 5.0 Å radius included HIS41, MET49, PRO52, TYR54, CYS145, HIS164, MET165, GLU166, LEU167, PRO168, ASP187, ARG188, GLN189, THR190, and GLN192. The conventional hydrogen bonding interactions between M^{Pro} and **3c** were observed, such as those between the H atom of the OH group in SER144 and the N atom of the CN group in **3c**, between the H atom of the NH₂ group in HIS163 and the N atom of the CN group in **3c**, between the S atom of the SH group in MET165 and the H atom of the NH₂ group in **3c**, and those between the H atom of the NH₂ group in THR190 and the N atom of the NH₂ group in **3c**. Additionally, one carbon hydrogen bond was formed with GLN189. Moreover, one π -Sulfur interaction was displayed with CYS145, and one π -Alkyl interaction was observed with MET49.

For **4-(2',6'-Dimethyl-4'-cyanophenoxy)-2-(5''-cyanopyridin-2''ylamino)quinoline (3d)**, the amino acid residues of M^{Pro} within a 5.0 Å radius included HIS41, MET49, PRO52, TYR54, CYS145, HIS164, MET165, GLU166, LEU167, PRO168, ASP187, ARG188, GLN189, THR190, and GLN192. The conventional hydrogen bonding interactions between M^{Pro} and **3d** were established, such as those involving the H atom of the NH₂ group in SER144 and the N atom of the CN group in **3d**, and those connecting the S atom of the SH group in MET165 with the H atom of the NH₂ group in **3d**. Additionally, one carbon hydrogen bond formed with GLN192. Two π -Sulfur interactions were interacted with CYS145 and MET165, and one π -Alkyl interaction occurred with MET49.

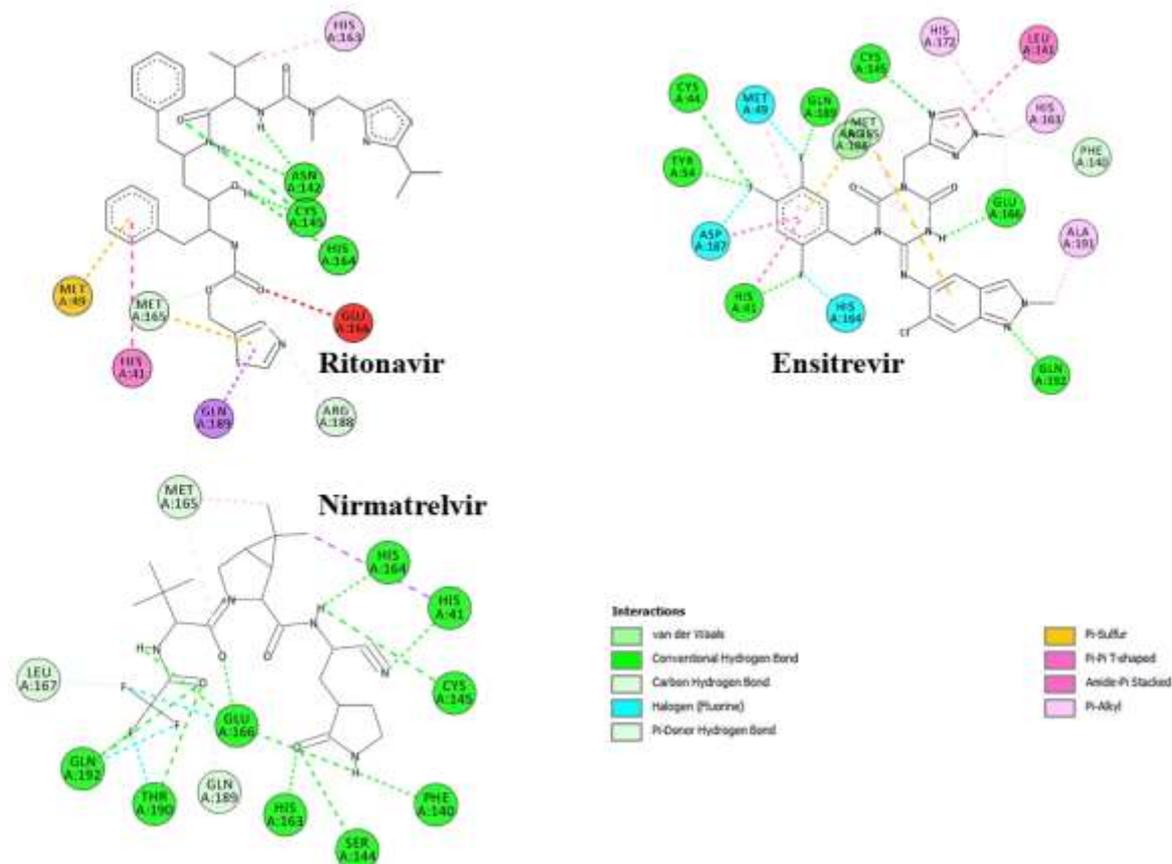


Figure S4 2-dimensional interaction diagram between commercially accessible medications and M^{Pro} .

For ritonavir, the amino acid residues of M^{Pro} within 5.0 Å included HIS41, MET49, TYR54, PHE140, LEU141, ASN142, GLY143, SER144, CYS145, HIS163, HIS164, MET165, GLU166, LEU167, PRO168, ASP187, ARG188, and GLN189. The conventional hydrogen bonding interactions between M^{Pro} and ritonavir were observed, such as those between the O-atom of the C=O group in ASN142 and the H-atom of the NH₂ group in ritonavir, between the S atom of the SH group in CYS145 and the H atom of OH group in ritonavir, between the O atom of the C=O group in HIS164 and the H atom of OH group in ritonavir. Two carbon hydrogen bonds were formed with MET165, ARG188 and one π -Donor interaction was displayed with GLU166. Furthermore, a π - π stacking interaction was noted between the HIS41 residue and the benzene ring of ritonavir.

For ensitrevir, the amino acid residues of M^{Pro} within a 5.0 Å radius surrounding ensitrevir included HIS41, MET49, TYR54, PHE140, LEU141, GLY143, SER144, CYS145, HIS163, HIS164, MET165, GLU166, HIS172, and ARG188. The conventional hydrogen bonding interactions between M^{Pro} and ensitrevir were demonstrated, such as those between the H atom of the NH₂ group in HIS41 and the F atom of the benzyl group in ensitrevir, between the H atom of the OH group in TYR54 and the F atom of the benzyl group in ensitrevir, between the H atom of the SH group in CYS145 and the N atom of the NH₂ group in ensitrevir, between the O atom of the C=O group in GLU166 and the H atom of the NH₂ group in ensitrevir, between the H atom of the NH₂ group in GLN192 and the N atom

of ensitrevir, and between the H atom of the NH₂ group in GLN189 and the F atom of the benzyl group in ensitrevir. Three carbon hydrogen bonds interacted with PHE140, MET165, and GLU166, and one π -donor hydrogen bond was formed with GLU166. Additionally, one π -Sulfur interaction occurred with MET165, and two π -Alkyl interactions were observed with HIS163 and HIS172. A π - π stacking interaction between the HIS41 residue of M^{PRO} and the benzene ring of ensitrevir was also observed.

For nirmatrelvir, the amino acid residues of M^{PRO} within a 5.0 Å radius included HIS41, MET49, PRO52, TYR54, HIS163, HIS164, MET165, GLU166, PRO168, ARG188, THR190, ALA191, and GLN192. Nirmatrelvir demonstrated the conventional hydrogen bonding interactions with M^{PRO}. These included interactions between the H atom of the NH₂ group in HIS41 and the N atom of the CN group in nirmatrelvir, the N atom of the NH₂ group in PHE140 and the H atom of the NH₂ group in nirmatrelvir, the H atom of the OH group in SER144 and the O atom of the C=O group in nirmatrelvir, the H atom of the SH group in CYS145 and the N atom of the NH₂ group in nirmatrelvir, the H atom of the NH₂ group in HIS163 and the O atom of the C=O group in nirmatrelvir, the O atom of the C=O group in HIS164 and the H atom of the NH₂ group in nirmatrelvir, the N atom of the NH₂ group in THR190 and the O atom of the C=O group in nirmatrelvir, and the N atom of the NH₂ group in GLN192 and the O atom of the C=O group in nirmatrelvir. Additionally, three conventional hydrogen bonding interactions formed between the O atom of the C=O group in GLU166 and the H atom of the NH₂ group in nirmatrelvir, between the O atom of the C=O group in GLU166 and the H atom of the NH₂ group in nirmatrelvir, and between the N atom of the NH₂ group in GLU166 and the O atom of the C=O group in nirmatrelvir. Two carbon hydrogen bonds were displayed with MET165 and GLN189.

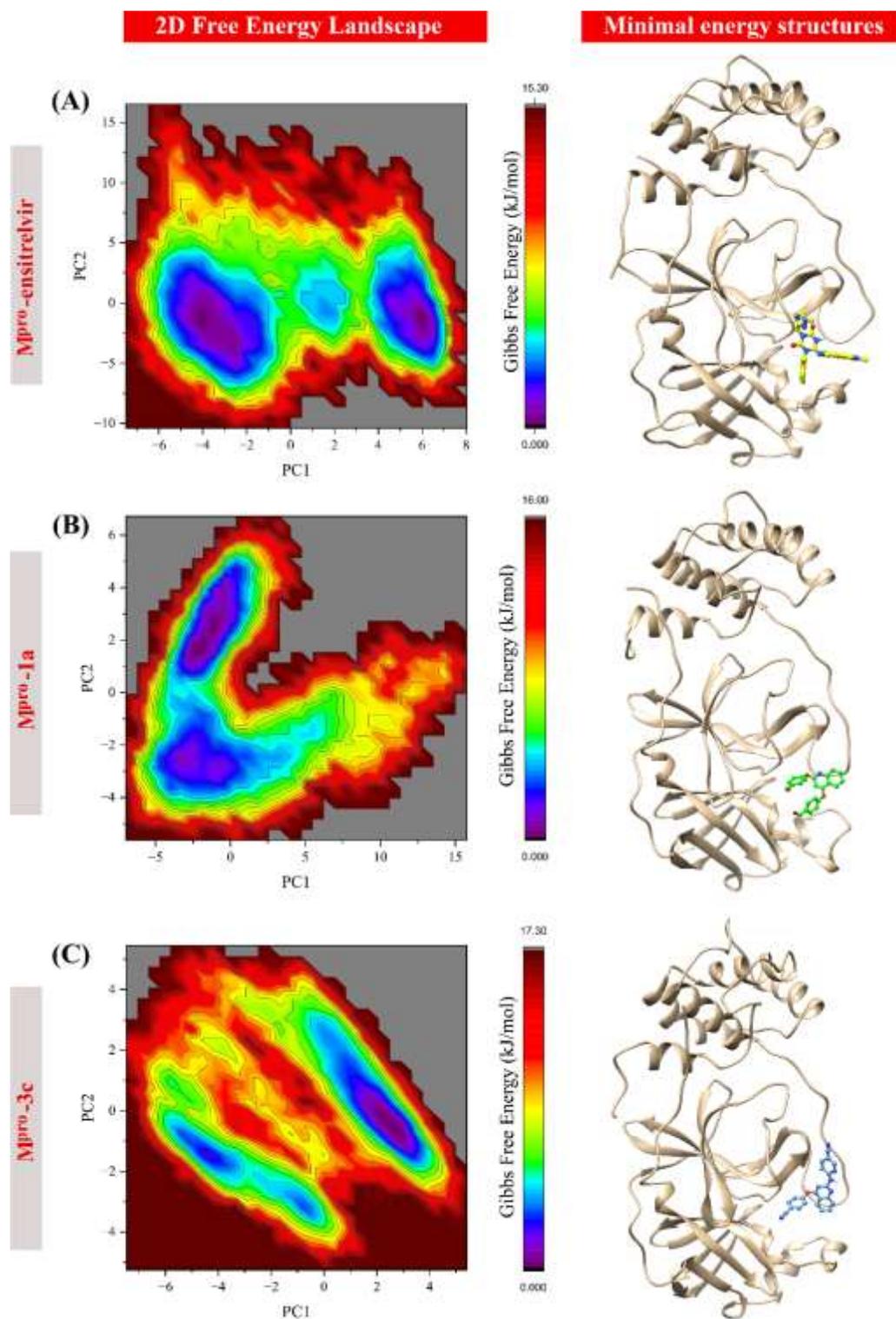


Figure S5 The Gibbs energy landscapes of (A) M^{pro}-nirmatrelvir, (B) M^{pro}-1a, and (C) M^{pro}-3c during a 300 ns MD simulation.

Table S1 %cells viability and vilucidal activity of quinoline derivatives **1-3(a-d)** and current medications (Ritonavir, Ensitrevir, and Nirmatrelvir).

Reagents	Concentration ($\mu\text{g/mL}$)	%Viability	Log Reduction	Efficacy
1a	1000	47.26 \pm 1.94	-	-
	500	73.93 \pm 2.20	2.890 \pm 0.086	99.87 \pm 0.03
	250	82.62 \pm 2.21	2.813 \pm 0.175	99.84 \pm 0.07
	125	83.81 \pm 1.94	2.784 \pm 0.105	99.83 \pm 0.04
	62.5	84.88 \pm 0.61	2.707 \pm 0.125	99.80 \pm 0.05
	31.25	85.48 \pm 0.89	2.681 \pm 0.088	99.77 \pm 0.04
	15.625	87.02 \pm 1.71	-	-
1b	1000	26.67 \pm 1.61	-	-
	500	81.67 \pm 1.21	2.918 \pm 0.055	99.88 \pm 0.02
	250	83.21 \pm 1.05	2.848 \pm 0.139	99.85 \pm 0.05
	125	84.17 \pm 2.07	2.828 \pm 0.113	99.85 \pm 0.04
	62.5	84.52 \pm 0.84	2.771 \pm 0.140	99.82 \pm 0.06
	31.25	86.67 \pm 0.89	2.763 \pm 0.139	99.82 \pm 0.06
	15.625	88.10 \pm 0.61	-	-
1c	1000	28.57 \pm 1.34	-	-
	500	71.43 \pm 1.46	3.032 \pm 0.156	99.90 \pm 0.03
	250	76.67 \pm 1.10	2.911 \pm 0.123	99.87 \pm 0.03
	125	79.76 \pm 0.34	2.848 \pm 0.139	99.85 \pm 0.05
	62.5	85.48 \pm 2.34	2.828 \pm 0.113	99.85 \pm 0.04
	31.25	86.79 \pm 0.29	2.729 \pm 0.132	99.81 \pm 0.06
	15.625	94.64 \pm 0.87	-	-
1d	1000	67.38 \pm 1.02	-	-
	500	83.10 \pm 2.11	3.063 \pm 0.075	99.91 \pm 0.01
	250	90.12 \pm 1.71	3.021 \pm 0.039	99.90 \pm 0.01
	125	91.19 \pm 1.61	2.926 \pm 0.050	99.88 \pm 0.01
	62.5	94.88 \pm 1.21	2.828 \pm 0.113	99.85 \pm 0.04
	31.25	97.86 \pm 1.05	2.729 \pm 0.132	99.81 \pm 0.06
	15.625	101.80 \pm 1.46	-	-
2a	1000	77.38 \pm 1.21	-	-
	500	96.98 \pm 1.98	2.996 \pm 0.096	99.90 \pm 0.02
	250	97.19 \pm 3.07	2.965 \pm 0.043	99.89 \pm 0.01
	125	97.71 \pm 2.45	2.848 \pm 0.139	99.85 \pm 0.05

Reagents	Concentration ($\mu\text{g/mL}$)	%Viability	Log Reduction	Efficacy
	62.5	98.75 \pm 1.54	2.828 \pm 0.113	99.85 \pm 0.04
	31.25	99.79 \pm 1.03	2.807 \pm 0.107	99.84 \pm 0.04
	15.625	104.38 \pm 1.16	-	-
2b	1000	54.18 \pm 1.78	-	-
	500	93.85 \pm 1.45	3.078 \pm 0.122	99.91 \pm 0.02
	250	98.23 \pm 1.88	3.000 \pm 0.003	99.90 \pm 0.00
	125	98.85 \pm 1.98	3.139 \pm 0.107	99.93 \pm 0.02
	62.5	100.83 \pm 0.74	2.792 \pm 0.103	99.83 \pm 0.04
	31.25	100.52 \pm 1.62	2.708 \pm 0.103	99.80 \pm 0.05
	15.625	101.88 \pm 1.12	-	-
2c	1000	25.71 \pm 1.62	-	-
	500	97.24 \pm 2.58	2.926 \pm 0.050	99.88 \pm 0.01
	250	97.59 \pm 2.76	2.841 \pm 0.140	99.85 \pm 0.05
	125	98.28 \pm 1.38	2.820 \pm 0.112	99.84 \pm 0.04
	62.5	98.51 \pm 0.86	2.707 \pm 0.125	99.80 \pm 0.05
	31.25	99.83 \pm 1.90	2.686 \pm 0.089	99.79 \pm 0.04
	15.625	102.86 \pm 0.77	-	-
2d	1000	37.62 \pm 0.61	-	-
	500	88.69 \pm 1.21	2.799 \pm 0.104	99.84 \pm 0.04
	250	95.18 \pm 0.54	2.707 \pm 0.125	99.80 \pm 0.05
	125	100.00 \pm 2.04	2.686 \pm 0.089	99.79 \pm 0.04
	62.5	100.71 \pm 1.07	2.686 \pm 0.089	99.79 \pm 0.04
	31.25	105.36 \pm 2.04	2.681 \pm 0.223	99.77 \pm 0.09
	15.625	109.17 \pm 1.11	-	-
3a	1000	36.89 \pm 1.10	-	-
	500	95.21 \pm 1.56	3.346 \pm 0.088	99.95 \pm 0.01
	250	97.60 \pm 0.29	3.000 \pm 0.003	99.90 \pm 0.00
	125	98.54 \pm 0.53	2.933 \pm 0.049	99.88 \pm 0.01
	62.5	99.69 \pm 1.17	2.890 \pm 0.086	99.87 \pm 0.03
	31.25	99.90 \pm 1.45	2.807 \pm 0.107	99.84 \pm 0.04
	15.625	102.92 \pm 1.37	-	-
3b	1000	38.56 \pm 0.68	-	-
	500	97.43 \pm 1.02	3.191 \pm 0.082	99.93 \pm 0.01
	250	97.90 \pm 1.28	3.160 \pm 0.075	99.93 \pm 0.01

Reagents	Concentration ($\mu\text{g/mL}$)	%Viability	Log Reduction	Efficacy
	125	98.67 ± 1.98	3.032 ± 0.066	99.91 ± 0.01
	62.5	102.19 ± 1.15	2.969 ± 0.059	99.89 ± 0.02
	31.25	103.33 ± 0.36	2.926 ± 0.050	99.88 ± 0.01
	15.625	80.10 ± 0.71	-	-
3c	1000	25.88 ± 1.77	-	-
	500	81.11 ± 1.81	3.782 ± 0.106	99.98 ± 0.00
	250	85.44 ± 2.38	3.452 ± 0.094	99.96 ± 0.01
	125	86.22 ± 0.57	3.160 ± 0.073	99.93 ± 0.01
	62.5	89.33 ± 0.47	3.032 ± 0.066	99.91 ± 0.01
	31.25	89.56 ± 0.68	2.965 ± 0.043	99.89 ± 0.01
	15.625	90.22 ± 1.10	-	-
3d	1000	50.11 ± 0.68	-	-
	500	90.22 ± 0.42	3.153 ± 0.071	99.93 ± 0.01
	250	91.22 ± 1.85	3.063 ± 0.075	99.91 ± 0.01
	125	92.67 ± 1.19	3.032 ± 0.066	99.91 ± 0.01
	62.5	92.89 ± 1.13	3.011 ± 0.186	99.90 ± 0.04
	31.25	94.89 ± 1.93	3.000 ± 0.003	99.90 ± 0.00
	15.625	97.44 ± 1.75	-	-
Ritonavir	1000	21.14 ± 0.62	-	-
	500	27.05 ± 0.82	-	-
	250	54.95 ± 0.82	-	-
	125	71.71 ± 1.82	3.578 ± 0.084	99.97 ± 0.01
	62.5	73.52 ± 1.85	3.503 ± 0.121	99.97 ± 0.01
	31.25	75.71 ± 1.15	3.284 ± 0.038	99.95 ± 0.00
	15.625	76.10 ± 0.49	3.246 ± 0.074	99.94 ± 0.01
	7.813	76.86 ± 2.57	3.153 ± 0.037	99.93 ± 0.01
Ensitrevir	1000	24.76 ± 1.32	-	-
	500	54.00 ± 0.84	-	-
	250	80.19 ± 1.64	3.261 ± 0.050	99.94 ± 0.01
	125	81.05 ± 0.94	3.106 ± 0.083	99.92 ± 0.02
	62.5	85.62 ± 2.78	3.000 ± 0.067	99.90 ± 0.02
	31.25	87.33 ± 0.59	3.021 ± 0.107	99.90 ± 0.03
	15.625	89.33 ± 1.17	3.000 ± 0.100	99.90 ± 0.02
Nirmatrelvir	1000	55.81 ± 1.10	-	-

Reagents	Concentration ($\mu\text{g/mL}$)	%Viability	Log Reduction	Efficacy
	500	73.51 ± 1.28	3.153 ± 0.037	99.93 ± 0.01
	250	75.25 ± 1.55	3.021 ± 0.039	99.90 ± 0.01
	125	77.15 ± 0.84	3.011 ± 0.186	99.90 ± 0.04
	62.5	78.18 ± 1.55	3.000 ± 0.003	99.90 ± 0.00
	31.25	80.65 ± 1.63	2.828 ± 0.113	99.85 ± 0.04
	15.625	82.86 ± 1.07	-	-
Paxlovid (Ritonavir/ Nirmatrelvir ; 1:3)	1000	42.28 ± 1.68	-	-
	500	80.10 ± 0.71	3.191 ± 0.082	99.93 ± 0.01
	250	83.14 ± 1.20	3.000 ± 0.100	99.90 ± 0.02
	125	85.81 ± 1.55	2.990 ± 0.082	99.90 ± 0.02
	62.5	86.76 ± 0.84	2.990 ± 0.082	99.90 ± 0.02
	31.25	92.76 ± 1.64	2.905 ± 0.082	99.87 ± 0.03
	15.625	92.76 ± 1.64	-	-

^aResults are expressed as mean \pm SD of quadruplicate experiments.

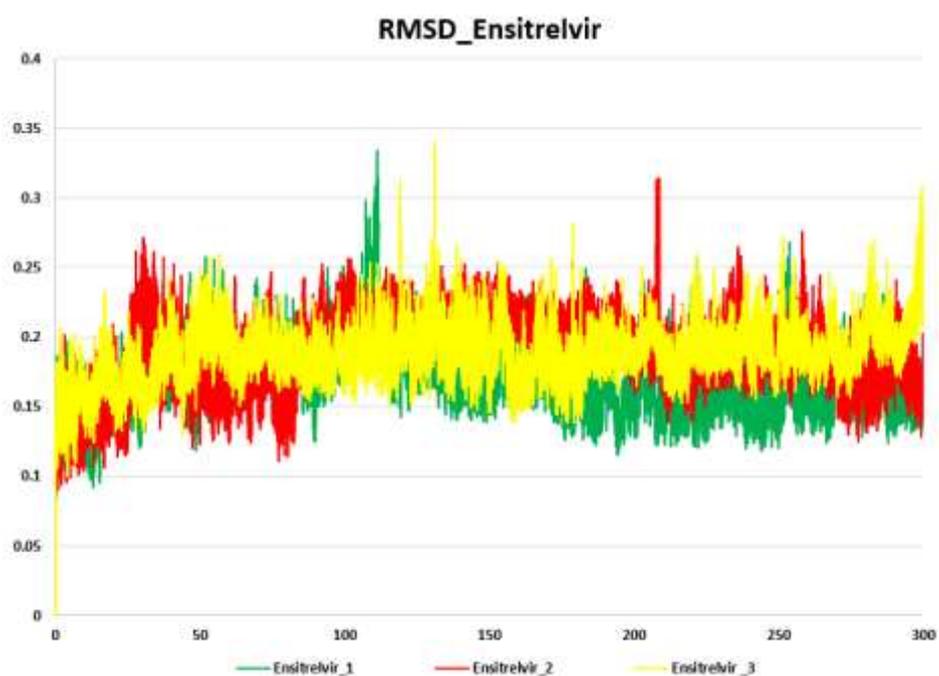


Figure S6 RMSD analysis of MD simulation triplicates of the M^{Pro}- ensitrelvir complex over 300 ns.

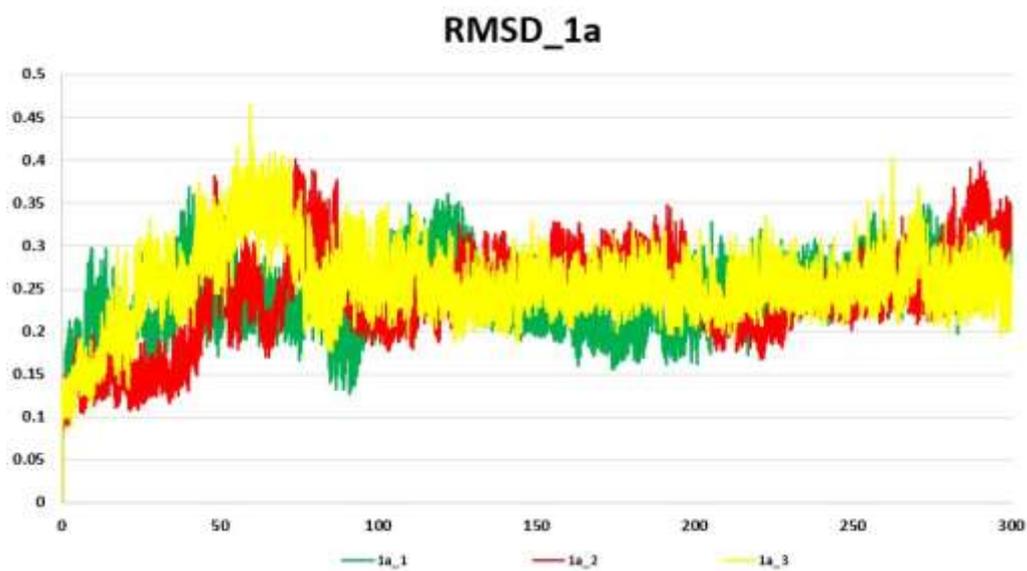


Figure S7 RMSD analysis of MD simulation triplicates of the M^{pro}-**1a** complex over 300 ns.

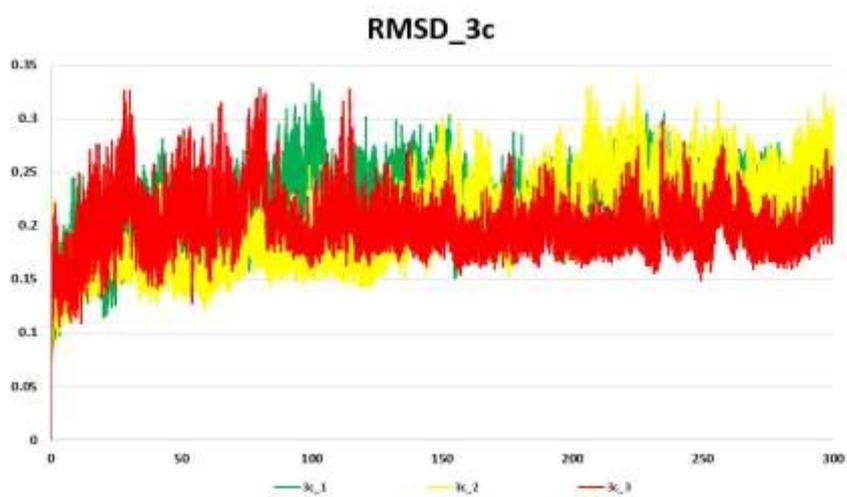


Figure S8 RMSD analysis of MD simulation triplicates of the M^{pro}-**3c** complex over 300 ns.

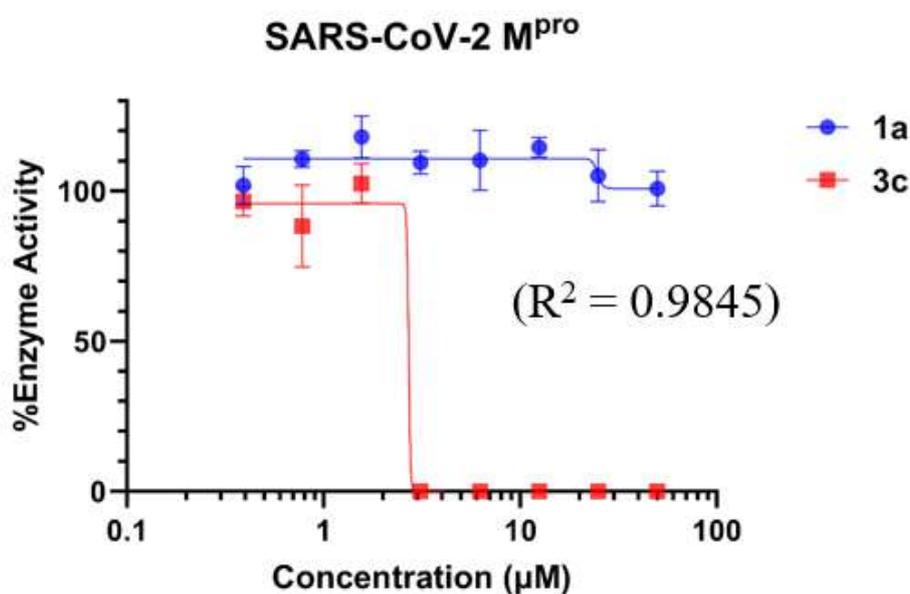


Figure S9 Dose-dependent curve of 1a and 3c against SARS-CoV-2 M^{pro} (R² = 0.9845).

Table S2 Binding energies and individual component energy values obtained from the MM-GBSA and MM-PBSA calculation of M^{pro}- nirmatrelvir, M^{pro}-ensitrelvir, M^{pro}-**1a** and M^{pro}-**3c** complexes.

Energetic terms (kcal/mol)	1a	3c	Ensitrelvir	Nirmatrelvir
ΔE_{vdw}	-37.99 ± 2.56	-40.06 ± 2.94	-50.50 ± 3.59	-52.11 ± 3.08
ΔE_{ele}	-2.24 ± 2.56	-3.49 ± 0.93	-8.57 ± 1.58	-13.28 ± 1.54
$\Delta G_{\text{polar/GB}}$	7.44 ± 0.87	7.57 ± 0.14	14.78 ± 1.29	18.87 ± 1.14
$\Delta G_{\text{non-polar/GB}}$	-4.61 ± 0.28	-4.88 ± 0.44	-5.76 ± 0.33	-6.32 ± 0.27
$\Delta G_{\text{polar/PB}}$	8.20 ± 0.86	8.19 ± 1.02	15.62 ± 1.57	19.38 ± 1.24
$\Delta G_{\text{non-polar/PB}}$	-3.34 ± 0.13	-3.69 ± 0.20	-4.50 ± 0.29	-4.49 ± 0.14
$-T\Delta S$	6.90 ± 0.14	6.54 ± 0.13	9.89 ± 0.13	6.12 ± 0.34
$\Delta G_{\text{bind/GB}}$	-30.51 ± 2.56	-34.32 ± 3.14	-40.16 ± 4.81	-48.74 ± 2.85
$\Delta G_{\text{bind/PB}}$	-28.47 ± 2.45	-32.51 ± 2.86	-38.05 ± 4.70	-46.40 ± 2.75

ΔE_{vdw} = van der Waals energy, ΔE_{elec} = electrostatic energy, $\Delta G_{\text{polar/GB}}$ = The polar solvation free energy obtained from the generalized Born method, $\Delta G_{\text{non-polar/GB}}$ = Nonpolar solvation free energy obtained from the generalized Born method, $\Delta G_{\text{polar/PB}}$ = The polar solvation free energy obtained from the Poisson-Boltzmann method, $\Delta G_{\text{non-polar/PB}}$ = Nonpolar solvation energy obtained from the Poisson-Boltzmann method, $-T\Delta S$ = Interaction entropy, $\Delta G_{\text{Binding/GB}}$ = Binding free energy (kcal/mol) from the generalized Born method, $\Delta G_{\text{Binding/PB}}$ = Binding free energy (kcal/mol) from the Poisson-Boltzmann method.

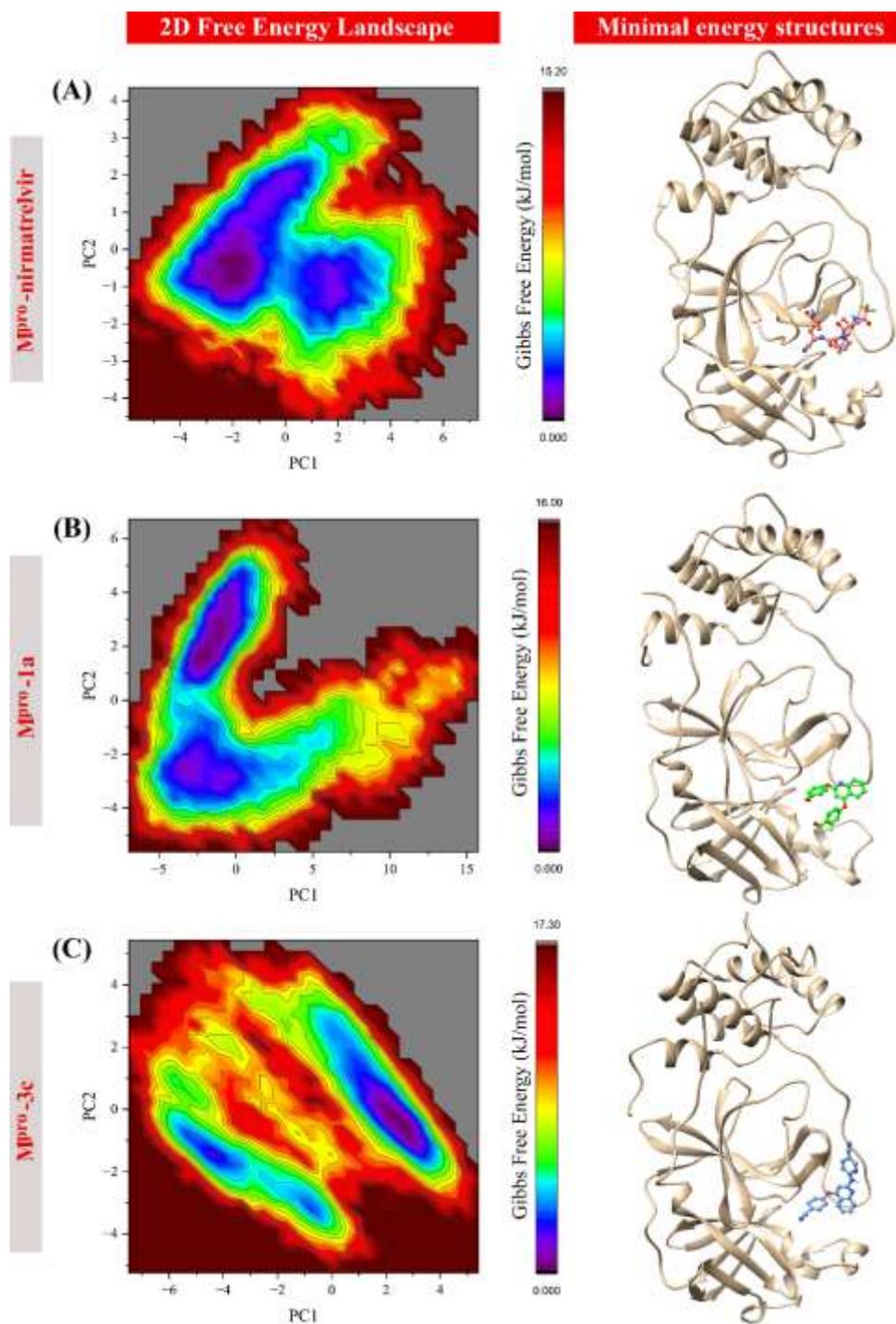


Figure S10 The Gibbs energy landscapes of (A) M^{pro}-nirmatrelvir, (B) M^{pro}-1a, and (C) M^{pro}-3c during a 300 ns MD simulation.

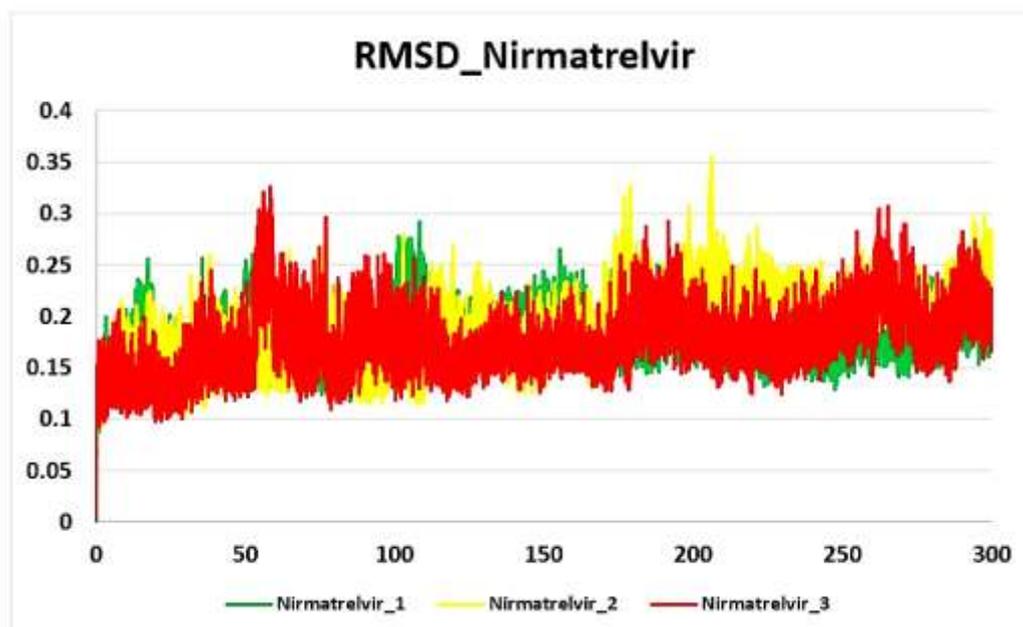


Figure S11 RMSD analysis of MD simulation triplicates of the M^{pro}-nirmatrelvir complex over 300 ns.